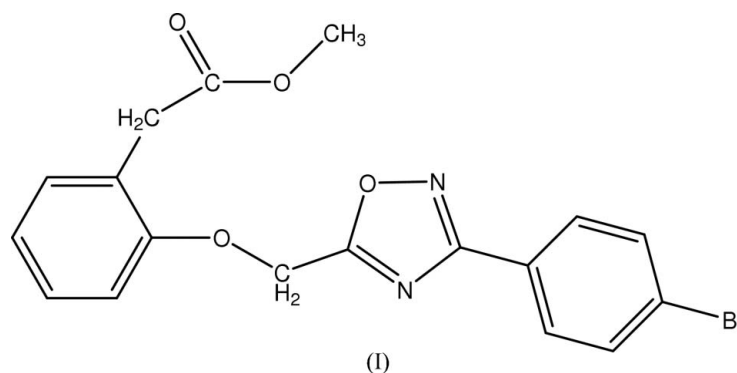


Zhi-Qian Liu,* Hai-Bo Wang,
Yue-Qing Pu and Xiao-Chen YanDepartment of Applied Chemistry, College of
Science, Nanjing University of Technology,
Xinmofan Road No.5 Nanjing, Nanjing 210009,
People's Republic of ChinaCorrespondence e-mail:
wanghaibo@njut.edu.cn

Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.008$ Å
 R factor = 0.046
 wR factor = 0.142
Data-to-parameter ratio = 13.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Methyl {2-[3-(4-bromophenyl)-1,2,4-oxa-
diazol-5-ylmethoxy]phenyl}acetateThe title compound, $\text{C}_{18}\text{H}_{15}\text{BrN}_2\text{O}_4$, was synthesized by the
reaction of methyl (2-hydroxyphenyl)acetate and 3-(4-
bromo)phenyl-5-chloromethyl-1,2,4-oxadiazole. Weak intra-
molecular $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds are observed in the
crystal structure.Received 19 October 2005
Accepted 14 February 2006

Comment

1,2,4-Oxadiazole derivatives are of great interest because of
their biological properties. Some derivatives of 1,2,4-
oxadiazole have intrinsic analgesic (Terashita *et al.*, 2002),
anti-inflammatory (Nicolaidis *et al.*, 1998) and anti-
picornaviral (Romero, 2001) properties and show high efficacy
as agonists [*e.g.* for muscarinic (Macor *et al.*, 1996), adrenergic
(Quagliato & Andrae, 2002) and 5-hydroxytryptamine (Gur
et al., 2001)] and antagonists [*e.g.* for angiotensin (Naka & Kubo,
1999) and adhesion (Jurazyk *et al.*, 1997)] for different
receptors. We report here the crystal structure of the title
compound, (I).The molecular structure of (I) is shown in Fig. 1 in which
dashed lines indicate intramolecular $\text{C}-\text{H}\cdots\text{N}$ hydrogen
bonds (Table 2). The bond lengths and angles are given in
Table 1.

Experimental

Methyl (2-hydroxyphenyl)acetate (20 mmol) was dissolved in
acetone (20 ml) and potassium carbonate (30 mmol) was added in
one portion. 3-(4-Bromophenyl)-5-chloromethyl-1,2,4-oxadiazole
(20 mmol) in acetone (20 ml) was added to this mixture. The resulting
mixture was refluxed for 6 h, then concentrated under reduced
pressure to afford crude (I). Pure (I) was obtained by recrystallization
from ethyl acetate. Crystals of (I) suitable for X-ray diffraction were
obtained by slow evaporation of an ethanol solution. ^1H NMR
(CDCl_3): δ 7.94–7.97 (*m*, 2H), 7.61–7.63 (*m*, 2H), 7.23–7.29 (*m*, 2H),
6.97–7.03 (*m*, 2H), 5.35 (*s*, 2H), 3.72 (*s*, 2H), 3.69 (*s*, 3H).

Crystal data

C₁₈H₁₅BrN₂O₄
M_r = 403.23
 Monoclinic, C2/c
a = 20.011 (4) Å
b = 9.643 (2) Å
c = 18.089 (4) Å
 β = 98.33 (3)°
V = 3453.7 (12) Å³
Z = 8

D_x = 1.551 Mg m⁻³
 Mo Kα radiation
 Cell parameters from 25 reflections
 θ = 9–12°
 μ = 2.41 mm⁻¹
T = 293 (2) K
 Block, colourless
 0.40 × 0.40 × 0.20 mm

Data collection

Nonius CAD-4 diffractometer
 ω/2θ scans
 Absorption correction: ψ scan
 (North *et al.*, 1968)
T_{min} = 0.446, *T_{max}* = 0.645
 3130 measured reflections
 3038 independent reflections
 1670 reflections with *I* > 2σ(*I*)

R_{int} = 0.030
 θ_{max} = 25.0°
h = 0 → 23
k = 0 → 11
l = -21 → 21
 3 standard reflections
 every 200 reflections
 intensity decay: none

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.046
wR(*F*²) = 0.142
S = 1.02
 3038 reflections
 227 parameters
 H-atom parameters constrained

w = 1/[σ²(*F_o*²) + (0.07*P*)² + 0.6*P*]
 where *P* = (*F_o*² + 2*F_c*²)/3
 (Δ/σ)_{max} = 0.001
 Δρ_{max} = 0.42 e Å⁻³
 Δρ_{min} = -0.36 e Å⁻³
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.0030 (3)

Table 1

Selected geometric parameters (Å, °).

| | | | |
|------------|-----------|-------------|-----------|
| Br—C16 | 1.898 (5) | N1—C11 | 1.287 (6) |
| O1—C2 | 1.333 (6) | N1—C12 | 1.375 (6) |
| O1—C1 | 1.441 (6) | N2—C12 | 1.310 (6) |
| O2—C2 | 1.187 (5) | C2—C3 | 1.496 (7) |
| O3—C9 | 1.382 (6) | C3—C4 | 1.499 (7) |
| O3—C10 | 1.405 (6) | C10—C11 | 1.474 (7) |
| O4—C11 | 1.333 (6) | C12—C13 | 1.467 (6) |
| O4—N2 | 1.416 (5) | | |
| C2—O1—C1 | 116.2 (4) | O3—C10—C11 | 108.8 (4) |
| C9—O3—C10 | 117.7 (4) | N1—C11—O4 | 113.8 (5) |
| C11—O4—N2 | 106.3 (3) | N1—C11—C10 | 128.0 (5) |
| C11—N1—C12 | 102.7 (4) | O4—C11—C10 | 118.2 (4) |
| C12—N2—O4 | 102.7 (4) | N2—C12—N1 | 114.5 (4) |
| O2—C2—O1 | 123.1 (5) | N2—C12—C13 | 122.1 (5) |
| O2—C2—C3 | 125.3 (5) | N1—C12—C13 | 123.5 (4) |
| O1—C2—C3 | 111.6 (4) | C18—C13—C12 | 122.0 (4) |
| C2—C3—C4 | 112.9 (4) | C14—C13—C12 | 119.0 (5) |
| O3—C9—C4 | 114.1 (4) | C15—C16—Br | 119.0 (4) |
| O3—C9—C8 | 124.6 (5) | C17—C16—Br | 120.5 (4) |

Table 2

Hydrogen-bond geometry (Å, °).

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|-------------------------|-------------|---------------|-----------------------|-------------------------|
| C14—H14A...N1 | 0.93 | 2.57 | 2.899 (7) | 101 |
| C18—H18A...N2 | 0.93 | 2.60 | 2.889 (7) | 100 |

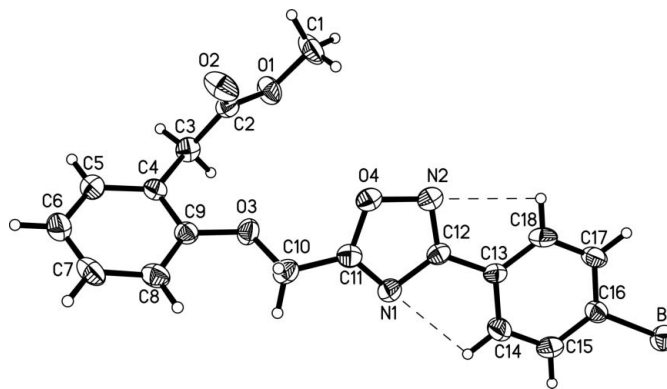


Figure 1

A view of the molecular structure of (I), with dashed lines indicating intramolecular C—H...N hydrogen bonds. Displacement ellipsoids are drawn at the 30% probability level

All H atoms were placed in calculated positions, with C—H distances in the range 0.93–0.97 Å. They were included in the riding-model approximation, with *U*_{iso}(H) = 1.2*U*_{eq}(C) or 1.5*U*_{eq}(methyl C).

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

References

- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Gur, E., Dremencov, E., Lerer, B. & Newman, M. E. (2001). *Eur. J. Pharmacol.* **411**, 115–122.
- Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.
- Juraszyk, H., Gante, J., Wurziger, H., Bernotat-Danielowski, S. & Melzer, G. (1997). Int. Patent Appl. No. WO199744333.
- Macor, J. E., Ordway, T., Smith, R. L., Verhoest, P. R. & Mack, R. A. (1996). *J. Org. Chem.* **61**, 3228–3229.
- Naka, T. & Kubo, K. (1999). *Curr. Pharm. Des.* **5**, 453–472.
- Nicolaides, D. N., Fylaktakidou, K. C., Litinas, K. E. & Hadjipavlou-Litina, D. (1998). *Eur. J. Med. Chem.* **33**, 715–724.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Quagliato, D. A. & Andrae, P. M. (2002). Int. Patent Appl. WO200206250.
- Romero, J. R. (2001). *Expert Opin. Invest. Drugs*, **10**, 369–379.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
- Siemens (1996). *SHELXTL*. Version 5.06. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Terashita, Z., Naruo, K. & Morimoto, S. (2002). Int. Patent Appl. WO2002060439.